

Available online at www.sciencedirect.com



Tetrahedron: *Asymmetry*

Tetrahedron: Asymmetry 18 (2007) 2787-2790

Enantioselective synthesis of phosphonate analogues of (*R*)- and (*S*)-homoserine

Dorota G. Piotrowska* and Iwona E. Głowacka

Bioorganic Chemistry Laboratory, Faculty of Pharmacy, Medical University of Łódź, 90-151 Łódź, Muszyńskiego 1, Poland

Received 4 October 2007; accepted 7 November 2007

Abstract—The enantioselective synthesis of (R)- and (S)-1-amino-3-hydroxypropylphosphonic acid, the phosphonate analogues of (S)- and (R)-homoserine, has been accomplished in four steps and good overall yield starting from diethyl (3R,5R)- or (3S,5S)-5-(hydroxy-methyl)-2-[(S)-1-phenylethyl]isoxazolidinyl-3-phosphonate, respectively. © 2007 Elsevier Ltd. All rights reserved.

9 2007 Elsevier Ltd. All fights festived.

1. Introduction

Homoserine 1 and homoserine lactone derivatives 2 have received considerable attention due to their biological activity (Fig. 1). While homoserine is an important amino acid involved in physiologically relevant transformations such as the biosynthesis of methionine and (S)-adenosylmethionine, its N-acylated lactones 2 (AHLs) play role as quorum-sensing signal molecules used by many Gram-negative bacteria to control their population density.^{1–5}



Figure 1. (S)-Homoserine 1 and N-acyl (S)-homoserine lactones 2.

Since chemical signalling processes used by bacteria have been recognised, the design and synthesis of new compounds which would be capable of inhibiting quorum-sensing activity and gene expression have become a challenging task. The inhibitory activity has been found, among others, for several derivatives of homoserine lactones and their structural analogues (Fig. 2).^{6–8}



Figure 2. Known QS inhibitors 3, 4 and 5.

So far, only a few syntheses of the racemic and enantiomerically pure phosphonate analogues of homoserine 6^{9-12} as well as oxaphospholanes **7a–7b** have been reported.^{13,14} Vasella employed *N*-glycosyl-*C*-dialkoxyphosphorylated nitrone in the highly selective multistep synthesis of (*R*)-phosphohomoserine (*R*)-6.⁹ However, this approach led to the formation of (*R*)-6 only. Moreover, this procedure suffers from several limitations when scaled-up. Roumestant et al. have obtained (*S*)-phosphohomoserine (*S*)-6 by the asymmetric alkylation of a Schiff base prepared from (1*S*,*2S*,*5S*)-2-hydroxypinan-3-one and diethyl aminomethylphosphonate.¹¹ Later, the same authors transformed the *N*-Cbz derivative of (*S*)-6 into the lactone (*S*)-7 using KF in the presence of 18-crown-6.¹⁴



^{*} Corresponding author. Tel.: +48 42 677 92 35; fax: +48 42 678 83 98; e-mail: dorota@ich.pharm.am.lodz.pl

^{0957-4166/\$ -} see front matter @ 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetasy.2007.11.007



Scheme 1. Retrosynthesis of phosphohomoserine 6.

Recently, the cycloaddition of (R)- and (S)-N-(1-phenylethyl)-C-phosphorylated nitrones **10** to allyl alcohol and further transformations of cycloadducts **9** into enantiomerically pure phosphonate analogues of 4-hydroxyprolines has been described.¹⁵ Since isoxazolidines (3R,5R,1'S)and (3S,5S,1'S)-**9** are easily available, we noticed their usefulness in the synthesis of both enantiomers of phosphohomoserine **6** (Scheme 1). Hydrogenolyses of isoxazolidines **9** in the presence Boc₂O would lead to *N*-Boc-aminodiols **8**, which could easily be transformed into (R)- and (S)-**6** via the oxidative cleavage of the terminal diol moiety followed by immediate NaBH₄ reduction of the aldehyde functional and hydrolysis of the resulted 1-amino-3-hydroxyphosphonates.

2. Results and discussion

Isoxazolidines (3R,5R,1'S)- and (3S,5S,1'S)-9 were synthesised according to the described procedure.¹⁵ They were subjected to catalytic hydrogenation performed in the presence of Boc₂O to give *N*-Boc-aminodiols (1R,3R)- and (1S,3S)-8 in good yields (74% and 75\%) after chromatographic purification (Scheme 2). Under these conditions, the cleavage of the N–O bond and simultaneous removal of the 1-phenylethyl substituent followed by the protection of the amino group were accomplished in a one-pot procedure. The ³¹P NMR spectra of 8 showed two signals in a 93:7 ratio due to the presence of the *tert*-butoxycarbonyl group. Treatment of diols (1R,3R)- and (1S,3S)-8 with sodium metaperiodate afforded aldehydes (*R*)- and (*S*)-11, which were immediately reduced with NaBH₄ to alcohols (*R*)- and (*S*)-12. The enantiomeric purities of (*R*)- and (S)-12 were confirmed by 31 P NMR analysis using quinine as an enantiodifferentiating agent. 16,17

The crude alcohols were contaminated (less then 10%) with the respective oxaphospholanes (2R,3R)- and (2S,3S)-13, due to partial cyclisation of the γ -hydroxypropylphosphonates 12 under basic conditions. Fortunately, oxaphospholanes 13 were easily removed chromatographically. although for the purpose of the syntheses of (R)- and (S)-6 their separation from the alcohol (R)- or (S)-12 is not necessary. The structure of (2R, 3R)-13 was established on the basis of the ¹H, ¹³C and ³¹P NMR spectra. This was further supported by the independent synthesis of (2S,3S)-13 from (2S,3S)-2-ethoxy-2-oxo-3- $\{N$ -hydroxy-N- $[(S)-1-phenylethyl]amino\}-1,2-oxaphospholane.^{18}$ Finally, phosphonates (R)- and (S)-12 were hydrolysed under acidic conditions and subsequently treated with propane oxide to give free α -aminophosphonic acids (R)- and (S)-6 in 95% and 94% yields, respectively.



In assigning the absolute configurations of *N*-Boc-aminodiols (1R,3R)- and (1S,3S)-**8**, we took advantage of the known configurations of starting isoxazolidines (3R,5R,1'S)- and (3S,5S,1'S)-**9**,¹⁵ since the transformations of (3R,5R,1'S)- and (3S,5S,1'S)-**9** into the respective (1R,3R)- and (1S,3S)-**8** did not involve any reactions at the stereogenic centres. From the diol (1R,3R)-**8** levorotatory phosphohomoserine (*R*)-**6** was obtained, which agrees



Scheme 2. Reagents and conditions: (a) Boc_2O , $H_2/Pd(OH)_2$, EtOH, rt, 24 h; (b) $NaIO_4$, $CH_2Cl_2-H_2O$, rt, 2 h; (c) $NaBH_4$, EtOH, 30 min; (d) 6 M HCl, 65 °C, 6 h.

2789

with the specific rotation determined previously for (R)-6 by Vasella.

3. Conclusions

A short and efficient synthesis of enantiomerically pure (R)- and (S)-phosphohomoserine **6**, from isoxazolidines (3R,5R,1'S)- and (3S,5S,1'S)-**9** has been reported. Hydrogenolysis of (3R,5R,1'S)- and (3S,5S,1'S)-**9** in the presence of Boc₂O gave the respective N-substituted 1-amino-3,5-dihydroxybutylphosphonates, (1R,3R)- and (1S,3S)-**8**, which were subsequently transformed into (R)- and (S)-**6** via reaction sequence consisted of standard NaIO₄ oxidation, NaBH₄ reduction and acid hydrolysis.

4. Experimental

¹H NMR spectra were recorded with a Varian Mercury-300 spectrometer; chemical shifts δ in ppm with respect to TMS; coupling constants J in Hz. ¹³C and ³¹P NMR spectra were recorded on a Mercury-300 machine at 75.5 and 121.5 MHz, respectively. IR spectra data were measured on a Infinity MI-60 FT-IR spectrometer. Melting points were determined in a capillary on a Electrothermal apparatus and are uncorrected. Elemental analyses were performed by the Microanalytical Laboratory of this Faculty on a Perkin Elmer PE 2400 CHNS analyser. Polarimetric measurements were conducted on a Perkin Elmer 241 MC apparatus. The following absorbents were used: column chromatography, Merck silica gel 60 (70– 230 mesh); analytical TLC plastic sheets silica gel 60 F₂₅₄.

4.1. *tert*-Butyl (1R,3R)- and (1S,3S)-1-(diethoxyphosphono)-3,4-dihydroxybutylcarbamates (1R,3R)- and (1S,3S)-8

4.1.1. tert-Butyl (1R,3R)-1-(diethoxyphosphono)-3,4-dihydroxybutylcarbamate (1R,3R)-8. A solution of isoxazolidine (3R, 5R, 1'S)-9 (0.400 g, 1.16 mmol) and Boc₂O (0.254 g, 1.16 mmol) in ethanol (2 mL) was kept under an atmospheric pressure of hydrogen over 20% Pd(OH)2-C (15 mg) at room temperature for 24 h. The suspension was filtrated through a layer of Celite. The solution was concentrated and the residue was chromatographed on a silica gel column with chloroform-methanol (first 100:1, 50:1 and later 10:1, v/v) to give (1R,3R)-8 (0.294 g, 74%) as a colourless oil. $[\alpha]_{D}^{20} = -2.7$ (*c* 1.3, CHCl₃). IR (film): $v = 3363, 2980, 2933, 1709, 1529, 1367, 1301, 1224, 1170, 1031, 973 cm⁻¹. ¹H NMR (CDCl₃): <math>\delta = 5.07$ (br d, J = 9.0 Hz, 1H, NH), 4.25–4.10 (m, 4H), 3.95–3.86 (very br m, 1H), 3.83 (br s, 1H), 3.64 (br dd, J = 10.8, 3.0 Hz, 1H), 3.52 (dd, J = 10.8, 6.3 Hz, 1H), 2.62 (br s, 1H), 2.08-1.90 (m, 1H), 1.90-1.72 (m, 2H), 1.45 (s, 9H), 1.34 (t, J = 6.9 Hz, 6H). ¹³C NMR (CDCl₃): $\delta = 155.0$ (d, J =6.6 Hz), 8.39, 69.40 (d, J = 10.6 Hz), 65.96, 63.30 (d, J = 6.8 Hz), 63.00 (d, J = 6.8 Hz), 44.64 (d, J = 156.3 Hz), 33.81 (d, J = 3.8 Hz), 28.53, 16.67 (d, 156.3 Hz), 33.81 (d, J = 3.8 Hz), 28.53, 16.67 (d, J = 7.5 Hz), 16.58 (d, J = 7.5 Hz). ³¹P NMR (CDCl₃): $\delta = 26.22$ (93%) and 25.63 (7%). Anal. Calcd for $C_{13}H_{28}NO_7P$: C, 45.74; H, 8.26; N, 4.10. Found: C, 46.01; H, 8.48; N, 4.36.

4.1.2. *tert*-Butyl (1*S*,3*S*)-1-(diethoxyphosphono)-3,4-dihydroxybutylcarbamate (1*S*,3*S*)-8. Isoxazolidine (3*S*,5*S*,1'*S*)-9 (0.479 g, 1.39 mmol) was hydrogenated as described in Section 4.1.1 to give (1*S*,3*S*)-8 (0.355 g, 75%) as a colourless oil. $[\alpha]_D^{20} = +2.8$ (*c* 1.2, CHCl₃). Anal. Calcd for C₁₃H₂₈NO₇P: C, 45.74; H, 8.26; N, 4.10. Found: C, 45.53; H, 8.55; N, 4.26.

4.2. *tert*-Butyl (*R*)- and (*S*)-1-(diethoxyphosphono)-3-hydroxypropylcarbamates (*R*)- and (*S*)-12

4.2.1. Synthesis of (R)-12. A mixture of N-Boc-aminodiol (1R,3R)-8 (0.128 g, 0.375 mmol), NaIO₄ (0.096 g, 0.450 mmol), water (2 mL) and methylene chloride (4 mL) was stirred at room temperature for 2 h. The organic layer was separated and the water phase was extracted with methylene chloride $(3 \times 5 \text{ mL})$. The organic extracts were collected, dried over MgSO₄ and evaporated in vacuo to afford crude (R)-11 (0.116 g, 100%), which was used immediately in the next step without purification. ¹H NMR (CDCl₃): $\delta = 9.74$ (dd, J = 2.7, 1.2 Hz, 1H, CHO), 4.96 (br d, J = 9.6 Hz, 1H, NH), 4.66–4.50 (m 1H, P– CH), 4.23-4.10 (m, 4H, $2 \times P-O-CH_2$), 2.90 (dddAB, $J_{AB} = 16.8 \text{ Hz}, J = 12.0, 4.8, 1.2 \text{ Hz}, 1\text{H}, HCH), 2.70$ (dddAB, J_{AB} = 16.8 Hz, J = 9.3, 8.4, 2.7 Hz, 1H, HCH), 1.44 (s, 9H, $(CH_3)_3C$), 1.34 (t, J = 6.9 Hz, 3H, P–O– CH₂CH₃), 1.33 (t, J = 6.9 Hz, 3H, P–O–CH₂CH₃). ³¹P NMR (CDCl₃): $\delta = 24.30$.

The crude aldehyde (*R*)-**11** (0.116 g, 0.375 mmol) was dissolved in ethanol (3 mL), cooled to 0 °C and NaBH₄ (0.028 g, 0.750 mmol) was added. After 30 min, the reaction mixture was concentrated, the residue was dissolved in chloroform (10 mL) and MgSO₄ (0.5 g) was added. The solids were removed, and the residue was evaporated and chromatographed on a silica gel column with chloroform-methanol (100:1, v/v). A less polar fraction (0.012 g, 10%) contained an impure (2*R*,3*R*)-**13**. In more polar fractions *N*-Boc-aminoalcohol (*R*)-**12** (0.100 g, 85%) was collected as a colourless oil.

4.2.1.1. *tert*-Butyl (*R*)-1-(diethoxyphosphono)-3-hydroxypropylcarbamate (*R*)-12. $[\alpha]_D^{20} = -15.3$ (*c* 1.0, CHCl₃). IR (film): $\nu = 3364$, 2980, 2932, 1706, 1530, 1367, 1231, 1169, 1050, 1028 cm⁻¹. ¹H NMR (CDCl₃): $\delta = 4.95$ (br dd, J = 10.2, 4.2 Hz, 1H, NH), 4.25-4.15 (m, 4H, $2 \times P-O-CH_2$), 3.80–3.60 (br m, 2H, CH_2 OH), 3.34 (dd, J = 8.7, 5.2 Hz, 1H, OH), 2.20–2.00 (m, 1H, P–CH), 1.80–1.60 (m, 2H, CH_2 CH₂OH), 1.45 (s, 9H, $(CH_3)_3$ C), 1.34 (t, J = 7.1 Hz, 6H, $2 \times P-O-CH_2CH_3$). ¹³C NMR (CDCl₃): $\delta = 156.59$ (d, J = 9.4 Hz), 80.87, 63.14 (d, J = 6.9 Hz), 62.81 (d, J = 6.6 Hz), 58.16 (d, J = 12.6 Hz), 44.18 (d, J = 157.5 Hz), 33.46, 28.49, 16.72 (d, J = 5.2 Hz), 16.65 (d, J = 5.4 Hz). ³¹P NMR (CDCl₃): $\delta = 26.74$. Anal. Calcd for C₁₂H₂₆NO₆P: C, 46.30; H, 8.42; N, 4.50. Found: C, 46.05; H, 8.68; N, 4.40.

4.2.1.2. *tert*-Butyl (2*R*,3*R*)-2-ethoxy-2-oxo-1,2-oxaphospholan-3-ylcarbamate (2*R*,3*R*)-13. Further purification of the less polar fraction on a silica gel column with chloroform-methanol (200:1, v/v) gave (2*R*,3*R*)-13 (0.010 g, 9%) as a colourless oil. $[\alpha]_{\rm D}^{20} = +32.6$ (*c* 0.7, CHCl₃). IR (film): v = 3268, 2980, 2933, 1741, 1708, 1530, 1455, 1367, 1256, 1167, 1044, 1006, 969, 830 cm⁻¹. ¹H NMR (CDCl₃): $\delta = 5.15$ (br s, 1H, NH), 4.29 (dddd, J = 13.8, 9.6, 6.8, 4.5 Hz, 1H), 4.20 (dq, J = 8.7, 7.2 Hz, 2H), 4.07 (dddd, J = 9.6, 8.4, 7.5, 5.7 Hz, 1H), 3.98 (br q, J = 6.9 Hz, 1H), 2.58 (ddddd, J = 24.9, 13.2, 7.5, 5.7, 4.5 Hz, 1H), 2.22 (ddq, J = 13.2, 8.4, 6.9 Hz, 1H), 1.45 (s, 9H), 1.37 (t, J = 7.2 Hz, 3H). ¹³C NMR (CDCl₃): $\delta = 155.65$ (d, J = 11.8 Hz), 80.55, 65.11 (d, J = 7.8 Hz), 63.85 (d, J = 6.6 Hz), 42.81 (d, J = 136.9 Hz), 32.69 (d, J = 10.9 Hz), 28.53, 16.69 (d, J = 5.7 Hz). ¹³P NMR (CDCl₃): $\delta = 41.01$. Anal. Calcd for C₁₀H₂₀NO₅P: C, 45.28; H, 7.60; N, 5.28. Found: C, 45.40; H, 7.85; N, 5.04.

4.2.2. *tert*-Butyl (*S*)-1-(diethoxyphosphono)-3-hydroxypropylcarbamate (*S*)-12. Following the procedure described in Section 4.2.1, from *N*-Boc-aminodiol (1*S*,3*S*)-8 (0.150 g, 0.440 mmol) *N*-Boc-aminoalcohol (*S*)-12 (0.109 g, 80%) was obtained. $[\alpha]_D^{20} = +15.2$ (*c* 1.1, CHCl₃). Anal. Calcd for C₁₂H₂₆NO₆P: C, 46.30; H, 8.42; N, 4.50. Found: C, 46.29; H, 8.21; N, 4.33.

4.3. (*R*)- and (*S*)-1-Amino-3-hydroxypropylphosphonic acids (*R*)- and (*S*)-6

4.3.1. (*R*)-1-Amino-3-hydroxypropylphosphonic acid (*R*)-**6.** A solution of *N*-Boc-aminoalcohol (*R*)-12 (0.060 g, 0.19 mmol) in 6 M HCl was refluxed for 6 h. The solvent was removed under reduced pressure, and the residue was dissolved in ethanol (1 mL) and neutralised with propylene oxide. The solvent was withdrawn with a pipette. The solid was washed with anhydrous ethanol and dried to afford (*R*)-**6** (0.029 g, 95%) as a white amorphous solid. Mp: 215–216 °C (lit.⁹ mp 214–217 °C). $[\alpha]_D^{20} = -6.6$ (*c* 1.1, H₂O) {lit.⁹ $[\alpha]_D^{25} = -6.2$ (*c* 1, H₂O)}. ¹H NMR (D₂O): $\delta = 3.82$ (t, J = 6.6 Hz, 2H, CH₂OH), 3.40 (ddd, J = 13.8, 8.7, 4.2 Hz, 1H, P–CH), 2.24–2.07 (m, 1H, P–CH–HCH), 2.05–1.85 (m, 1H, P–CH–HCH). ³¹P NMR (D₂O): $\delta = 14.01$. Anal. Calcd for C₃H₁₀NO₄P: C, 23.23; H, 6.50; N, 9.03. Found: C, 22.96; H, 6.31; N, 8.73.

4.3.2. (*S*)-1-Amino-3-hydroxypropylphosphonic acid (*S*)-**6.** Following the procedure described in Section 4.3.1, from *N*-Boc-aminoalcohol (*S*)-**6** (0.060 g, 0.19 mmol) the acid (*S*)-**12** (0.028 g, 94%) was obtained. Mp: 215–215.5 °C (lit.¹¹ mp 214 °C). $[\alpha]_D^{20} = +6.9$ (*c* 1.8, H₂O) {lit.¹¹ $[\alpha]_D = +7.3$ (*c* 1, H₂O)}. Anal. Calcd for C₃H₁₀NO₄P: C, 23.23; H, 6.50; N, 9.03. Found: C, 22.93; H, 6.45; N, 8.91.

4.4. Estimation of ee's of (R)-8 and (S)-8

A solution of (*R*)-8 (0.014 g, 0.040 mmol) in CDCl₃ (0.7 mL) containing quinine (0.075 g, 0.23 mmol) was analysed by ³¹P NMR spectroscopy. ³¹P NMR (CDCl₃): $\delta = 26.79$.

In a similar manner a solution of (S)-8 and quinine in CDCl₃ was analysed. ³¹P NMR (CDCl₃): $\delta = 26.59$.

Both solutions of (*R*)-8 and (*S*)-8 containing quinine used above were combined and the ³¹P NMR spectrum showed two signals at $\delta^{31}P = 26.79 [(R)-8], 26.59 [(S)-8].$

Acknowledgements

The authors wish to express their gratitude to Professor Andrzej E. Wróblewski for his interest and helpful suggestions during the preparation of this manuscript. We thank Mrs. Jolanta Płocka and Małgorzata Pluskota for their skilled experimental contributions. This work was supported by the Medical University of Łódź internal funds (502-13-332 and 503-3014-1).

References

- 1. Fuqua, C.; Greenberg, E. P. Nat. Rev. Mol. Cell. Biol. 2002, 3, 685–695.
- Lawrence, R. N.; Dunn, W. R.; Bycroft, B. W.; Camara, M.; Chhabra, S. R.; Williams, P.; Wilson, V. G. Br. J. Pharmacol. 1999, 128, 845–848.
- Chhabra, S. R.; Harty, C.; Hooi, D. S. W.; Daykin, M.; Williams, P.; Telford, G.; Pritchard, D. I.; Bycroft, B. W. J. Med. Chem. 2003, 46, 97–104.
- Park, S.-Y.; Hwang, B.-J.; Shin, M.-H.; Kim, J.-A.; Kim, H.-K.; Lee, J.-K. FEMS Microbiol. Lett. 2006, 261, 102–108.
- 5. Wang, Y.-J.; Leadbetter, J. R. Appl. Environ. Microbiol. 2005, 71, 1291–1299.
- Persson, T.; Hansen, T. H.; Rasmussen, T. B.; Skindersø, M. E.; Givskov, M.; Nielsen, J. Org. Biomol. Chem. 2005, 3, 253– 262.
- Mansfield, M.; Rasmussen, T. B.; Henzter, M.; Anderson, J. B.; Steinberg, P.; Kjelleberg, S.; Givskov, M. *Microbiology* 2002, 148, 1119–1127.
- Geske, G. D.; Wezeman, R. J.; Siegel, A. P.; Blackwell, H. E. J. Am. Chem. Soc. 2005, 127, 12762–12763.
- Vasella, A.; Voeffray, R. Helv. Chim. Acta 1982, 65, 1953– 1964.
- Chollet-Gravey, A.-M.; Vo-Quang, L.; Vo-Quang, Y.; Le Goffic, F. Synth. Commun. 1991, 21, 1847–1858.
- Quazzani, F.; Roumestant, M.-L.; Viallefont, P.; El Hallaoui, A. *Tetrahedron: Asymmetry* 1991, 2, 913–917.
- Alferov, K. V.; Zhukov, Y. N.; Khurs, E. N.; Osipova, T. I.; Khomutov, R. M. Russ. Chem. Bull., Int. Ed. 2001, 50, 316– 318.
- 13. Chollet-Gravey, A.-M.; Vo-Quang, L.; Vo-Quang, Y.; Le Goffic, F. Synth. Commun. 1993, 23, 561–569.
- Ryglowski, A.; Roumestant, M. L.; Viallefont, P. Synth. Commun. 1996, 26, 1739–1746.
- 15. Piotrowska, D. G.; Głowacka, I. E. *Tetrahedron: Asymmetry* **2007**, *18*, 1351–1363.
- Uccello-Barretta, G.; Pini, D.; Mastantuono, A.; Salvadori, P. Tetrahedron: Asymmetry 1995, 6, 1965–1975.
- Żymańczyk-Duda, E.; Skwarczyński, M.; Lejczak, B.; Kafarski, P. Tetrahedron: Asymmetry 1996, 7, 1277–1280.
- 18. Piotrowska, D. G. *Tetrahedron: Asymmetry*, submitted for publication.